

TORREFACTION OF OIL PALM FRONDS (OPF) AS A POTENTIAL FEEDSTOCK FOR ENERGY PRODUCTION PROCESSES

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Abstract. Oil palm fronds (OPF) and trunks contribute the highest biomass availability compared with other oil palm wastes. At the moment, they are usually left on the ground around the plantation area to decompose naturally and fertilize the soil. Previous researchers have focused on torrefaction of wood residues and other agricultural biomass with less attention has been paid to the utilization of Malaysia's biomass such as OPE. Therefore, in this study, torrefaction of OPF was conducted in a tubular reactor at temperatures between 200 and 300 °C and residence time of 30 min. The results reveal an improved heating value as the temperature was increased, from 16.81 to 20.32 MJ/kg after the torrefaction process. The van Krevelen diagram also proved that torrefaction OPF could be classified as an intermediate, between raw OPF and coal. This proves the potential of OPF as one of the alternative feedstocks for energy production process through torrefaction.

Keywords: torrefaction, biomass, Oil palm frond (OPF), waste management technology, renewable energy.

Introduction

Renewable biomass can be used as a fuel to reduce the dependency on fossil fuel for the production of heat, steam, and electricity for residential, industrial, and agricultural use. Replacing fossil fuels with biomass is more beneficial as it may reduce pollution and global warming. The palm oil industry is growing so fast and Malaysia is the second largest palm oil producer with estimated wastes of around 80 million dry tons (Umar, Jennings, & Ume, 2014). Furthermore, oil palm trees are being replanted to ensure sustainable oil productions, replacing low yields of ageing trees (Kamalrudin & Abdullah, 2014). Specifically, in Malaysia, Malaysia Palm Oil Board (MPOB) has even introduces Replanting Incentive Scheme to encourage oil palm tree replantation (Wahid & Simeh, 2010), warranting the continuous production of oil palm, and its biomass wastes.

The palm oil industry generates an abundance of biomass wastes in which only 10% of the tree is recovered as oil, leaving 90% as biomass waste in the forms of empty fruit bunch (EFB), palm kernel shell (PKS), oil palm frond (OPF), trunks, fiber, and palm oil mill effluent (POME) (Sukiran, Abnisa, Daud, Bakar, & Loh, 2017). As of July 2018, Malaysia has produced 10 million tonnes of

crude palm oil (CPO) whereby 45 million tonnes of fresh fruit bunch (FFB) was processed (MPOB, 2018). However, OPF is produced from the pruning of the palm trees in the plantation, and is produced continuously, generating up to 52 million tonnes of OPF per year (Sukiran et al., 2017). These wastes can be further utilized and developed as a one of the renewable energy sources which simultaneously minimizes the environmental problems and pollution. The current application of OPF includes roughage source in animal feed (Zahari, Abu Hassan, Wong, & Liang, 2003), raw material for biocomposites (Shinoj, Visvanathan, Panigrahi, & Kochubabu, 2011), converted to medium density fiber board (MDF), plywood or lumber for furniture manufacturing (Sukiran et al., 2017) while its juice has also been demonstrated as a source of carbohydrate, hence a sustainable carbon source (Tan, Jahim, Harun, Wu, & Mumtaz, 2016).

Oil palm biomass is found to have high potential energy value and its continuous availability can be utilized for renewable energy production for Malaysia. Due to its physical characteristics, oil palm biomass can be a useful feedstock to generate energy. One of the methods to convert biomass into useful feedstock is through torrefaction process. Torrefaction is a biomass thermal pretreatment technique at temperatures ranging from 200 to 300 °C

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(Rousset, Macedo, Commandré, & Moreira, 2012). Torrefaction is important to improve the biomass quality and reduce some of its problem such as low heating value and energy density, high moisture content (MC), low combustion efficiency, and high grinding energy requirements. Torrefaction increases the energy density of biomass by reducing its oxygen content leading to a solid biomass product with heating values nearing that of coal (Chen, Chua, & Lee, 2017). Torrefaction produces three products which are solid product, acidic aqueous products, and non-condensable gaseous products. The main product of the torrefaction process is the solid product which is also called as torrefied biomass. Torrefaction process can be divided into light torrefaction (200–240 °C) and severe torrefaction (260–300 °C) (Sabil, Aziz, Lal, & Uemura, 2013). Torrefied biomass is able to retain most of its chemical energy and can be ground easily compared with those of its original form. In addition, torrefied biomass has increased uniformity and improved quality in terms of its physical and chemical properties (Barta-Rajnai et al., 2016). The torrefied solids may be applied for energy production processes such as combustion, gasification, or production of bio-oil through pyrolysis (Louwes et al., 2017).

Torrefaction process is highly affected by the temperature in which the biomass is subjected to. This is true for various types of biomass such as mesocarp fiber and palm kernel in densified form (Mohd Faizal et al., 2018) and original form (Sabil et al., 2013), EFB (Abdul Rasid & Yusoff, 2017), and bamboo (M. F. Li, X. Li, Bian, & Chen, 2015). These studies mainly found that the increase of temperature decreased the percentage of mass and energy yields (Li et al., 2015; Abdul Rasid & Yusoff, 2017). Mass and energy yield are the significant indicators to determine the optimum parameter while considering also the higher heating value (HHV). Besides, when the calorific value increases, the moisture content and volatile matter decrease, with increased temperature and residence time (Sabil et al., 2013; Mohd Faizal et al., 2018). This is mainly due to the increase in devolatilization rate and decomposition of hemicellulose and cellulose when temperature increases.

However, the mass yield could be reduced too much at severe temperature, and longer residence time due to carbonization, dehydrogenation, and deoxygenation (Zhang et al., 2018). It was also observed that the color of the biomass changed from light brown to dark brown and dark black due to the structural modification of lignin and the formation of more chromophoric groups (Li et al., 2015) which may be related to light carbonization (Abdul Rasid & Yusoff, 2017). Severe temperature plays a more significant role in torrefied biomass properties, due to the decomposition of its lignocellulosic components, which are cellulose, hemicellulose as well as lignin, which becomes more significant, causing a significant weight loss of the biomass samples.

Previous researchers have focused on torrefaction forest residues (Macedo, Commandré, Rousset, Valette, & Pétrissans, 2018; Doddapaneni et al., 2017). Malaysia has the potential in the utilization of its own agricultural

biomass as they have been receiving little or no attention, especially oil palm wastes such as OPF. OPF is a byproduct of the cultivation of oil palm trees. OPF is either utilized or discarded at plantations and has not been optimally used. OPF contributes 14.71% (97 million tons per year) from the total oil palm wastes on a wet basis (Guangul, Sulaiman, & Ramli, 2012). They are usually left on the ground to decompose and fertilize the soil. There are limited studies on OPF torrefaction available in the literature, although studies on gasification have been done (Guangul et al., 2012). Hence, it is the aim of the present study to investigate the effect of torrefaction temperature on the characteristics of OPF as a potential renewable energy source through the torrefaction process.

1. Research methodology

1.1. Materials and analysis

OPF was obtained from Felda Lepar Hilir, Gambang, Kuantan, Pahang, Malaysia in 2016, where the average oil palm trees is approximately 10 years. Coal sample was obtained from TNB Research for comparison purposes. OPF sample was oven-dried at 105 °C for 24 h to provide a basis of the tested materials and to determine the oven-dry mass before torrefaction. After drying, the samples were ground and sieved into consistent size distribution of 250–500 µm. Then, the dried samples were transferred into labelled air-tight containers that contained silica gel to control MC and the samples were stored until further use. The ultimate analysis provides the composition of carbon (C), hydrogen (H), nitrogen (N), sulfur (S) using Perkin Elmer 2400 Series, in Central Lab, UMP. The oxygen (O) element was calculated by difference. The raw OPF and coal properties are shown in Table 1.

Table 1. Properties of raw OPF and coal

| | Ultimate analysis (wt.%) | | | | | HHV (MJ/kg) |
|---------|--------------------------|------|------|------|-------|----------------|
| | C | H | N | S | O | |
| Raw OPF | 42.26 | 7.03 | 2.81 | 0.22 | 47.68 | 16.81 |
| Coal | 63.35 | 5.81 | 1.21 | 0.25 | 29.39 | 26.03 |

1.2. Torrefaction experiment

The torrefaction of biomass was carried out using a vertical tubular stainless steel reactor as shown in Figure 1. A prescribed amount of biomass wastes (8.0 g) was weighed and placed in the reactor. The nitrogen (N₂) gas was supplied from a cylinder tank at 30 mL/min. The sample was heated up to the desired temperatures at a heating rate of 10 °C/min and once the desired temperature was achieved the reaction was allowed to take place for 30 min. The biomass was cooled down to room temperature before the torrefied biomass was retrieved and weighed. The torrefaction experiment was replicated at least three times to ensure consistency.

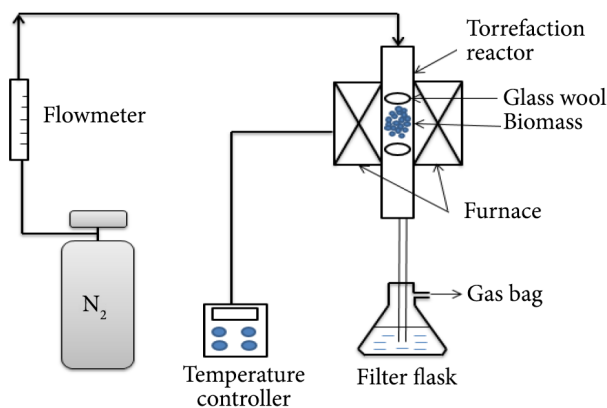


Figure 1. Schematic diagram of the experimental setup

1.3. Measurements

The mass yield, energy yield, and HHV were calculated according to Equations (1), (2), and (3) (Uemura, Omar, Tsutsui, & Yusup, 2011). Mass yield can be defined as the percentage of biomass solid retained after torrefaction process (g) divided by the initial mass before torrefaction (g) as shown in Equation (1):

$$y_m = \frac{m_T}{m_O} \times 100\%, \quad (1)$$

where y_m is the mass yield (%), m_O and m_T are the initial and final masses of solid (g), respectively. Energy yield indicates the total energy preserved in the torrefied biomass as shown in Equation (2):

$$y_E = y_m \times \frac{HHV_T}{HHV_O}, \quad (2)$$

where y_E is the energy yield (%), and HHV_T and HHV_O are the higher heating values of torrefied and initial biomass, respectively. The HHV value is defined as the amount of heat released by a specified quantity (initially at 25 °C) once it is combusted and the products have returned to an initial temperature as shown in Equation (3):

$$HHV = 0.4373Z_c - 1.6701, \quad (3)$$

where Z_c is the weight percentage (%) of elemental carbon. In addition, energy density is the amount of energy stored in a given system or region of space per unit volume as shown in Equation (4):

$$E_D = \frac{y_E}{y_m} \times 100\%, \quad (4)$$

where E_D is the energy density.

2. Results and discussions

2.1. Mass and energy yield

Mass yield can be defined as the percentage of biomass solid retained after torrefaction process (Equation (1)).

The mass yield of the torrefied OPF at different temperature is shown in Figure 2. Torrefaction process can be divided into light torrefaction (performed at 200, 220, and 240 °C) and severe torrefaction (performed at 260, 280, and 300 °C) (Sabil et al., 2013). The percentage of mass yield decreased with the increase in the operating temperature. This trend is consistent with the findings in the literature (Samad, Jamin, & Saleh, 2017; Mohd Faizal et al., 2018; Conag, Villahermosa, Cabatingan, & Go, 2018). The highest mass yield was 91.65% for the torrefaction at 200 °C, the mildest temperature. Meanwhile, the lowest mass yield was 68.96% obtained from the torrefaction process at 300 °C. Light torrefaction did not show any significant difference of mass yield for all temperatures.

From the viewpoint of mass yield, light torrefaction showed insignificant effect because of a little decrement from its original weight. The mass yield of light torrefaction was around 89–92% with an overall mass loss of 8–11%. The little decrement in light torrefaction might be due to the drying process occurred at below 220 °C. The decreasing of mass yield could be related to the proximate analysis where the MC and volatile matters decrease when the temperature increases (Wahid, Saleh, & Samad, 2017).

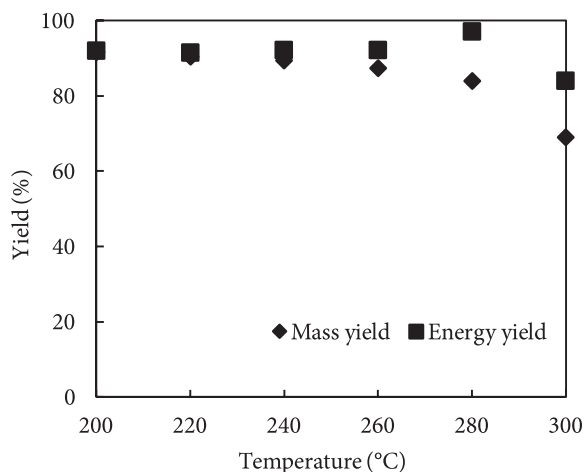


Figure 2. Mass and energy yield

Energy yield is important to measure torrefied biomass in terms of its energy density (Equation (2)). Energy density is defined as the amount of chemical energy stored in fuel per unit volume and if the energy density is more than unity then the desired energy gain can be achieved (Cellatoglu & Ilkan, 2015). Energy yield percentages at different temperatures for OPF are shown to have a decreasing trend as shown in Figure 2. It could be easily noticed that the torrefaction at 280 °C gave a higher energy yield compared to those performed at other temperatures. The decrement of energy yield may be caused by further decomposition of hemicellulose and cellulose in the biomass (Sabil et al., 2013). The observation result is consistent with previous works (Rodriguez, Martin-Lara, Blazquez, & Perez, 2017; Samad et al., 2017).

The energy density profile of OPF is shown in Figure 3. The energy density increased significantly at the temperature exceeding 240 °C and remained rather constant at 300 °C. It could be seen that for all temperatures, the energy density is greater than one of energy density, which suggests that the energy gain is achieved for torrefaction process based on definition of energy density. For OPF, the highest relative energy gain was at 300 °C. The energy density increases against temperature due to the increase of calorific value which is important to decrease the volume of biomass as a feedstock especially in biomass co-firing plant, as the torrefied biomass properties is similar to coal (Chen et al., 2017).

2.2. Ultimate analysis

The elemental analysis in terms of C, H, N, S, and O of the raw and torrefied biomass OPF is shown in Figure 4. Torrefaction increased the C content in the biomass along with the increasing temperature. The opposite was observed for H and O contents where both seemed to decrease with the increasing temperature. Previous researchers had also reported the same trend for C, H, and O contents (Iroba, Baik, & Tabil, 2017; Recari et al., 2017). The increment of the C content for OPF was around 19% at 300 °C compared with that of the raw OPF. These results indicate that torrefaction has a positive impact on the C content. The decrement of H and O content is mainly due to the dehydration of the biomass (Prins, Ptasinski, & Janssen, 2006). On the other hand, the increase in sulfur content is mainly due to the decreased for both the oxygen and hydrogen content at higher temperature, that essentially balances the weight percentages of the elements.

The changes in composition resulted in a decrease of O/C ratio. In general, raw OPF contains a low carbon content (43%) and high oxygen content (48%). However, coal contains a high carbon content (63%) and low oxygen content (30%). Figure 5 shows the O/C ratios of the torrefied biomass at different temperatures. The torrefaction process had positive influence to reduce the O/C ratio especially at more severe torrefaction temperature. The O/C ratio decreased by approximately 70%. The decrement of O/C ratio implies that there will be less smoke and vapor formation as well as reduction in energy loss during subsequent combustion and gasification processes (Tumulu, Sokhansanj, Wright, & Boardman, 2011).

To compare the OPF results with coal, a van Krevelen diagram was plotted as shown in Figure 6. The van Krevelen diagram is the graph of O/C ratio versus H/C ratio. As temperature increased, the O/C and H/C ratio are closer to those of coal. Before torrefaction, raw OPF had an O/C ratio of 1.13 and an H/C ratio of 0.17. At 300 °C, these O/C and H/C ratios were reduced to 0.67 and 0.11, respectively. Torrefied OPF could be classified as an intermediate product between raw OPF and coal. This is due to drying and devolatilization processes in torrefaction that remove surface water and carbon dioxide (Prins et al.,

2006). The van Krevelen diagram proved that the torrefied OPF can potentially be used as a feedstock for combustion and gasification since the physical properties are almost as similar as coal, as shown by lower and improved O/C and H/C ratio. This is especially evident at 300 °C.

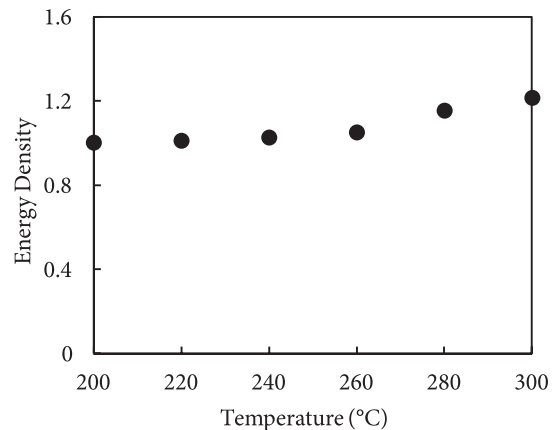


Figure 3. Energy density

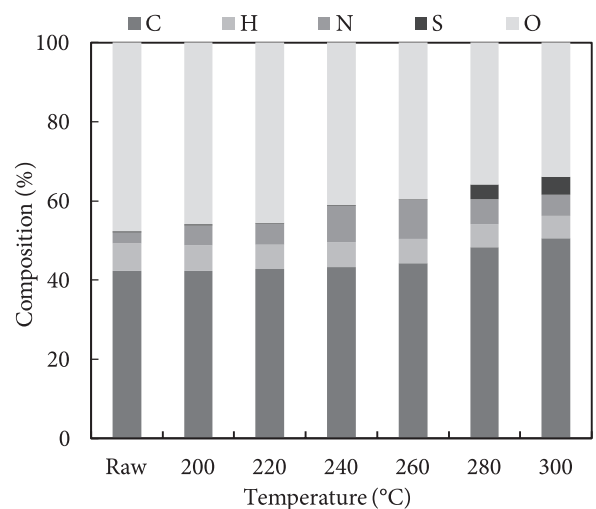


Figure 4. Ultimate analysis for OPF

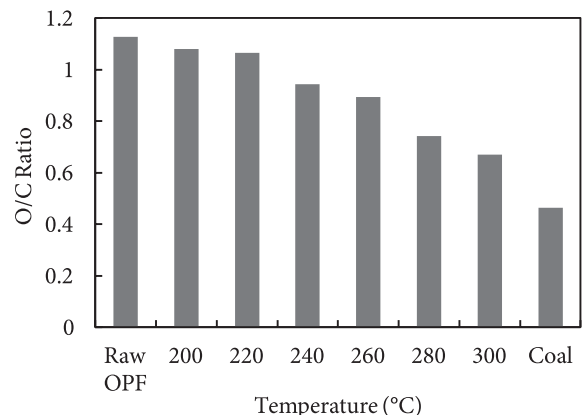


Figure 5. Effect of temperature on O/C ratio

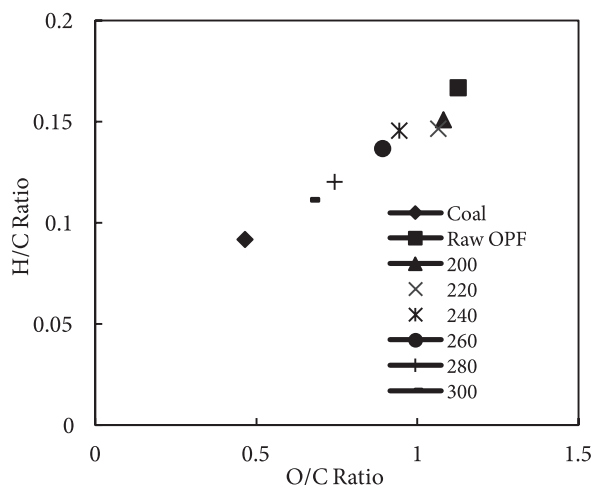


Figure 6. van Krevelen diagram of OPF

2.3. Proximate analysis

Table 2 presents the values of MC, volatiles matter (VM), fixed carbon (FC), and ash content (AC) for all samples before and after torrefaction. When the temperature was increased, MC and VM decreased, while the opposite was observed for FC and AC. The MC of torrefied OPF decreased significantly to 4.40% compared with that of the raw OPF (6.86%), which can be related to the decrease of mass yield. These results are in agreement with a previous study (Recari et al., 2017).

The percentages of moisture loss for each biomass were different for each biomass; bagasse (81–89%) and water hyacinth (83–89%) were higher than rice husks (70–84%), sawdust (71–82%), and peanut husks (71–79%). For gasification application, high MC is not favorable as it can reduce the feedstock heating value and in turn requires more energy for an efficient gasification process. However, MC around 20–35% can be considered for gasification depending on the type of gasifiers used (Bridgwater, Toft, & Brammer, 2002).

Table 2. Proximate analysis

| T (°C) | MC | VM | FC | AC |
|---------|------|-------|-------|-------|
| Raw OPF | 6.86 | 79.22 | 0.24 | 13.68 |
| 200 | 6.54 | 78.46 | 4.36 | 10.64 |
| 220 | 5.99 | 72.56 | 9.31 | 12.14 |
| 240 | 5.37 | 71.74 | 9.93 | 12.95 |
| 260 | 5.40 | 67.30 | 11.39 | 15.91 |
| 280 | 5.35 | 65.48 | 10.54 | 18.93 |
| 300 | 4.71 | 63.41 | 9.79 | 22.09 |
| Coal | 8.54 | 39.69 | 38.62 | 13.15 |

The VM content decreased from 79.22% to 63.41% across the temperature range, while the FC content increased from 3.23% to 12.01%. These trends are consistent with the compositional analysis results published by other researchers (Rodriguez et al., 2017; Conag et al., 2018).

High VM content is not favorable during gasification or combustion as it can cause excessive volatilization that can contribute to the formation of tar, soot, and smoke that affect the efficiency of gasification (Van der Stelt, Gerhauser, Kiel, & Ptasinski, 2011). The decrease in the MC and VM contents solved the storage problems of raw samples, where the reactivity of the torrefied biomass is reduced in storage (Tumulu et al., 2011). An increase in FC is suitable for further use in energy production processes as high FC can also increase the char formation during gasification process (Van der Stelt et al., 2011), which may increase the CO production during gasification. The AC increased steadily from 8.68% to 21.15%. Light torrefaction resulted in lower AC values compared with that of severe torrefaction, although sometimes high AC is needed for biomass gasification to induce the catalytic effect which may cause the cracking of high molecular weight compounds formed during gasification and change gasification dynamics (Fahmi, Bridgwater, Donnison, & Yates, 2008).

Fuel ratio is the ratio of FC content to VM content and can be used to evaluate the combustibility of coal and biochar (Huang, Suang, Chiueh, & Lo, 2017). Figure 7 illustrates the fuel ratio of raw and torrefied OPF where the values are all lower than 0.2. The fuel ratio for the torrefied OPF at 260 °C was the highest at 0.169 while at 300 °C, it was 0.154. The fuel ratio of coal on the other hand was 0.973. The issue here is that OPF could be an alternative renewable energy source but in terms of quality, OPF still far away than coal on the basis of fuel ratio. A way to improve the fuel ratio may be through combining the torrefied OPF with coal for co-combustion and co-gasification applications. The oil palm wastes can still be utilized due to its availability and at the same time can minimize environmental problems caused by coal. Further investigation in the future needs to be done to further clarify this.

2.4. Higher Heating Value (HHV)

Direct use of raw biomass as a fuel is inefficient due to its low calorific value, as it reduces the efficiency of the process. Therefore, torrefaction process increases the calorific

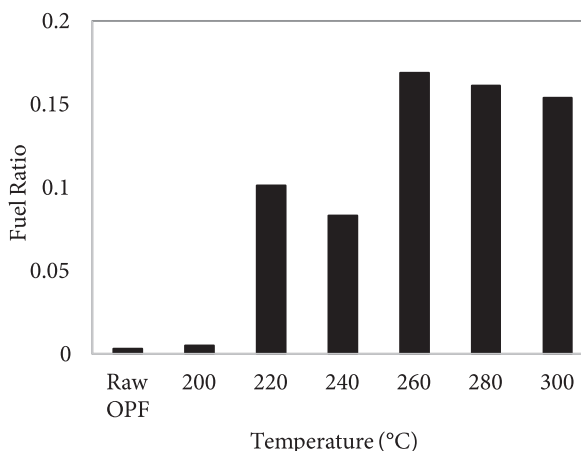


Figure 7. Fuel ratio of OPF

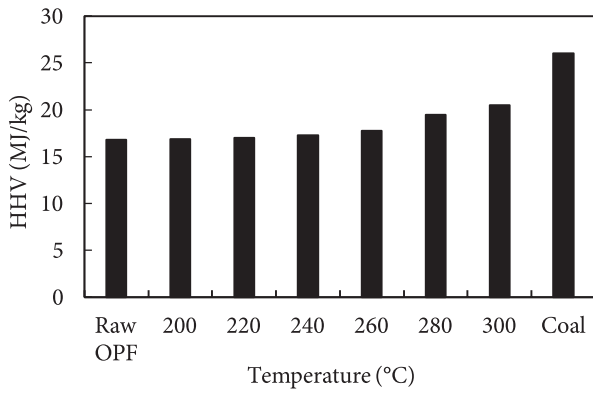


Figure 8. Higher heating value (HHV) for raw and torrefied OPF compared to coal

value as a function of temperature. From Figure 8, the HHV increased with the increasing torrefaction temperature. This result seems to be in a good agreement with the literatures (Irawan, Latifah Upe, & Meity Dwi, 2017; Cai, Fivga, Kaario, & Liu, 2017). At a temperature below 260 °C, the effect of temperature on the HHV is not as significant compared with that at higher temperature. This could be related to the increase on FC in the torrefied OPF (Mohd Faizal et al., 2018). Overall, the increase of the HHV of the torrefied OPF is about 0.34–21.70%. It is also important to highlight that the HHV of torrefied OPF is almost similar to coal especially at 300 °C, which indicates its suitability for industrial applications, such as in combustion and gasification.

Table 3. Surface structure of torrefied OPF

| | 500× | 1000× |
|--------|------|-------|
| Raw | | |
| 200 °C | | |
| 260 °C | | |
| 300 °C | | |

2.5. Surface morphology

Table 3 shows the changes on the surface structure of raw and torrefied OPFs under 500× and 1000× magnifications. The presence of horizontal lines was due to the fibrous structure of the OPF. At 200 °C, pores were clearly observed at both 500× and 1000× magnifications. These pores were present because they are fibrous materials. These observations were in a good agreement with the literature (Sabil et al., 2013). The changes of surface structure were more obvious at 260 °C. The pores disappeared at this temperature as observed at 1000× magnification because the fiber structure disassembled that might be due to decomposition. At 300 °C, the OPF structure degraded and in turn changed their internal structure. This is due to the decomposition temperature of hemicellulose, as OPF started to decompose at lower temperature of 150 °C. Meanwhile, at 300 °C severe decomposition occurred on the surface structure of OPF. On the basis of this observation, the decomposition temperature of hemicellulose can be said to affect the internal structure of OPF.

Conclusions

In this work, the characteristics of OPF as a potential energy source have been studied.

- It was found that the increase in torrefaction temperature leads to a reduction in mass and energy yields. The HHV of the torrefied OPF increased with increasing temperature. The most optimum temperature to achieve this is 300 °C.
- The results were also compared to raw OPF and coal which revealed that the characteristics of the torrefied OPF at more severe torrefaction temperature are almost similar to coal characteristics.
- The van Krevelen diagram proved that the torrefied OPF had the characteristics to allow it to be classified as an intermediate product between raw OPF and coal.
- This indicates that the torrefied OPF can be potentially used as a renewable and alternative feedstock for energy production processes such as the gasification and combustion application since the physical properties are almost similar with coal physical properties.

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